

Preparing Esters using Acid Anhydrides

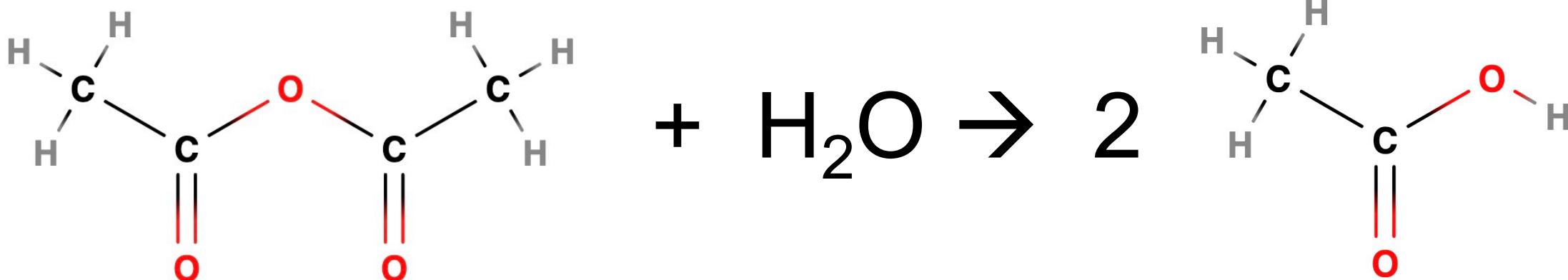
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Acid anhydrides in ester formation

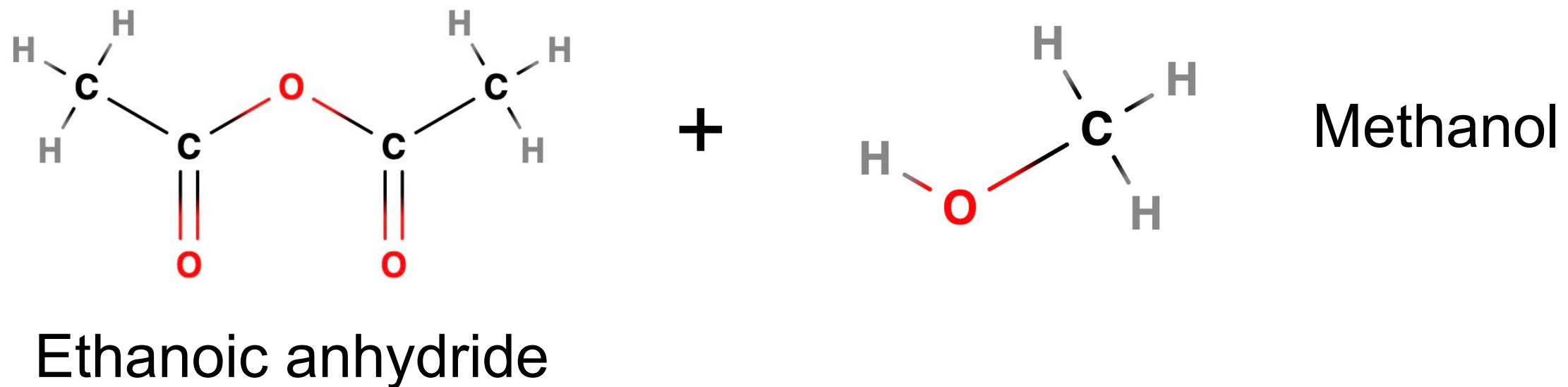
None reversible reaction.

Acid anhydrides will react with phenols to form esters.

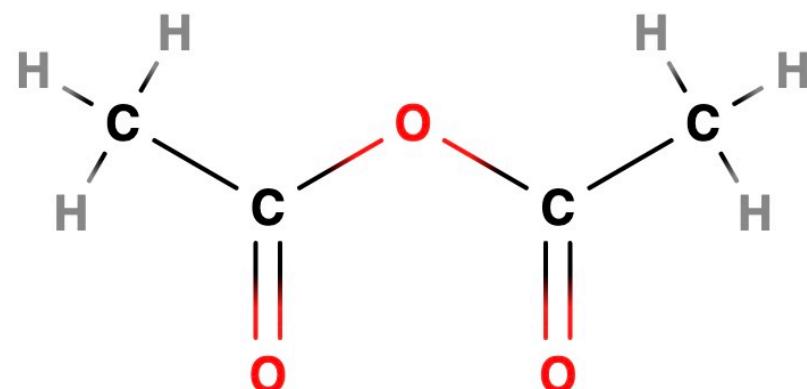
Completed under dry conditions – no water! Otherwise the acid anhydride reacts to form the acid.



Acid anhydrides in ester formation

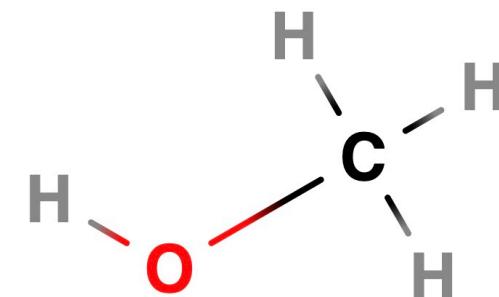


Acid anhydrides in ester formation

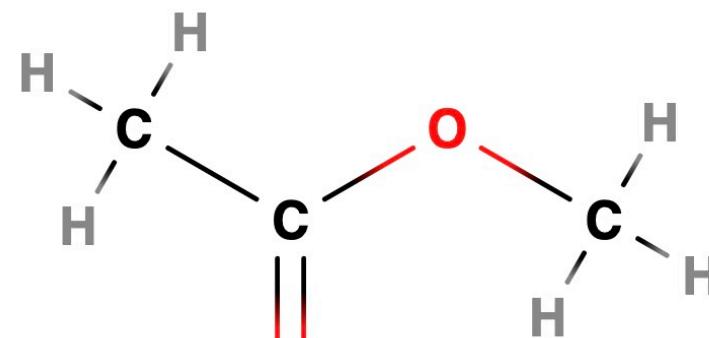


Ethanoic anhydride

+

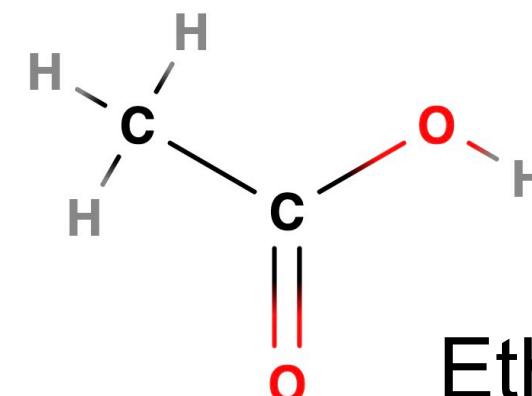


Methanol



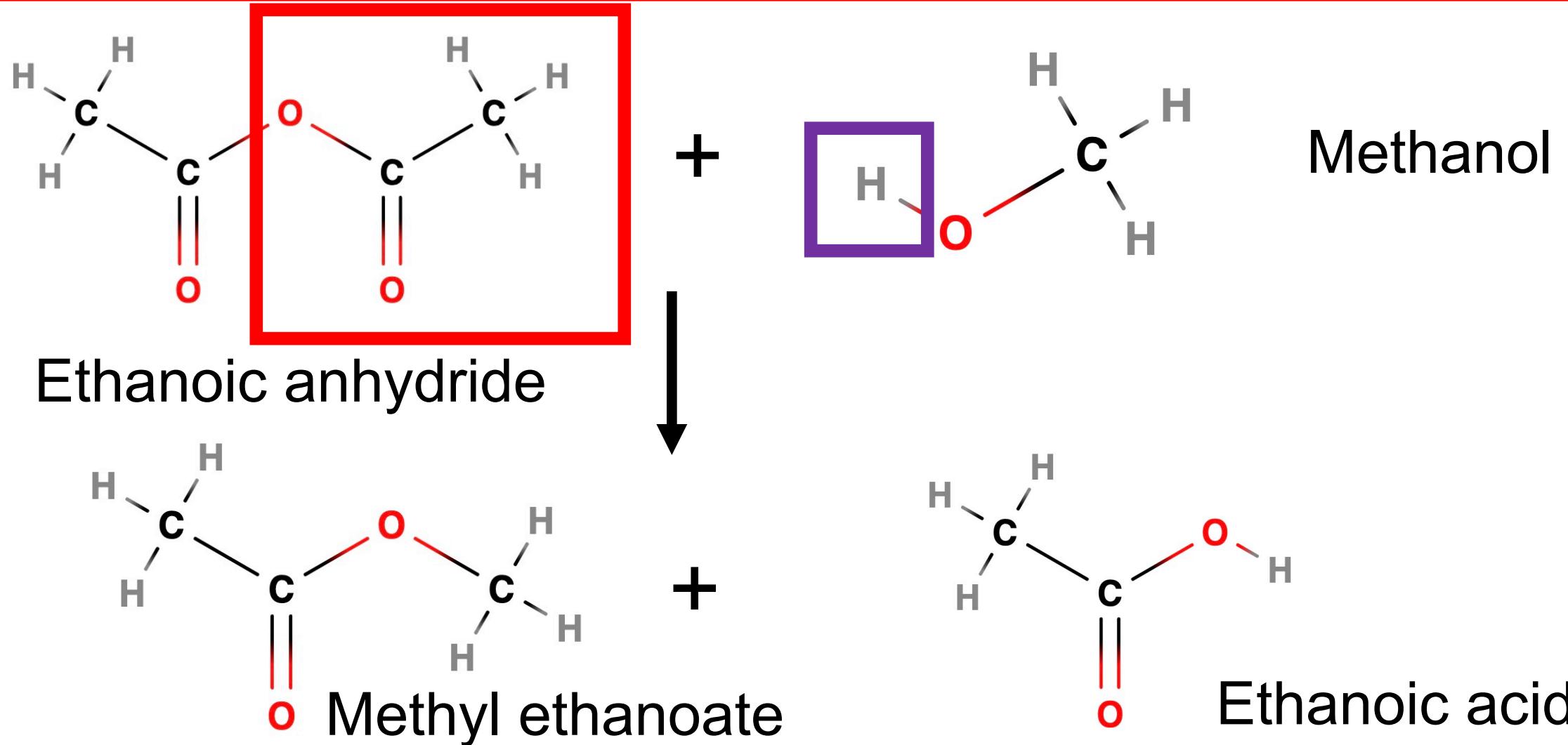
Methyl ethanoate

+

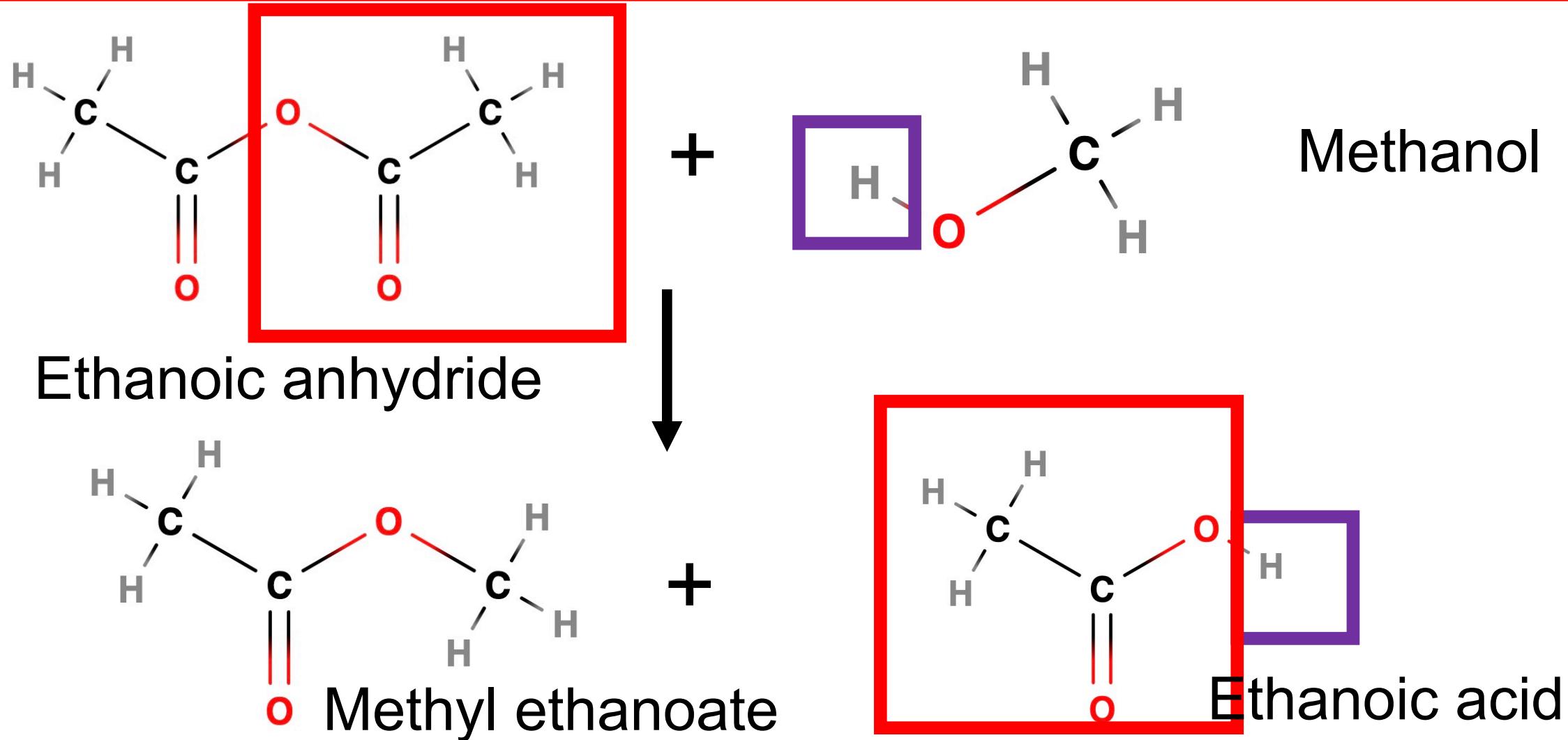


Ethanoic acid

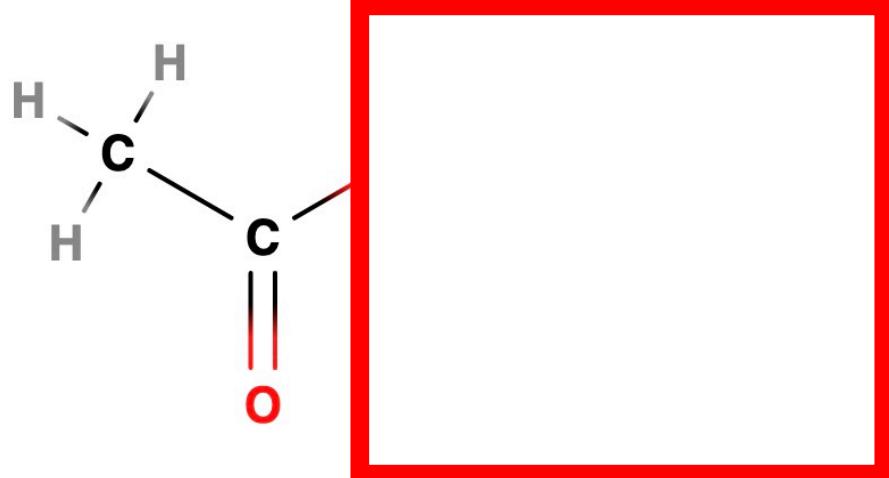
Acid anhydrides in ester formation



Acid anhydrides in ester formation

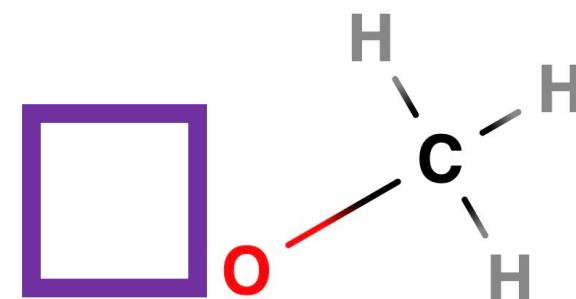


Acid anhydrides in ester formation

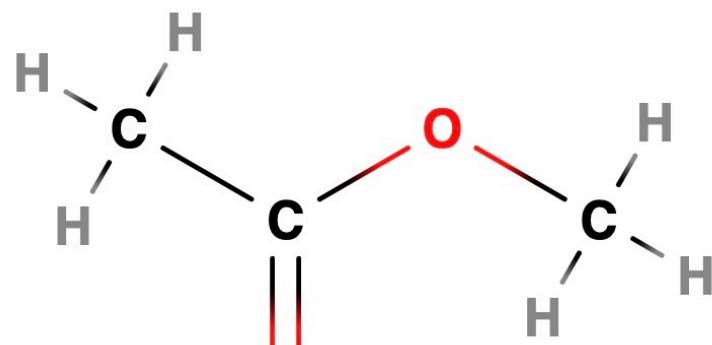


Ethanoic anhydride

+

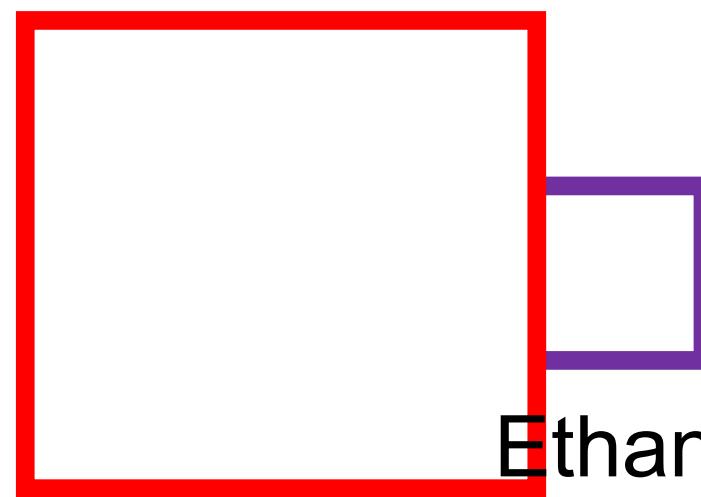


Methanol



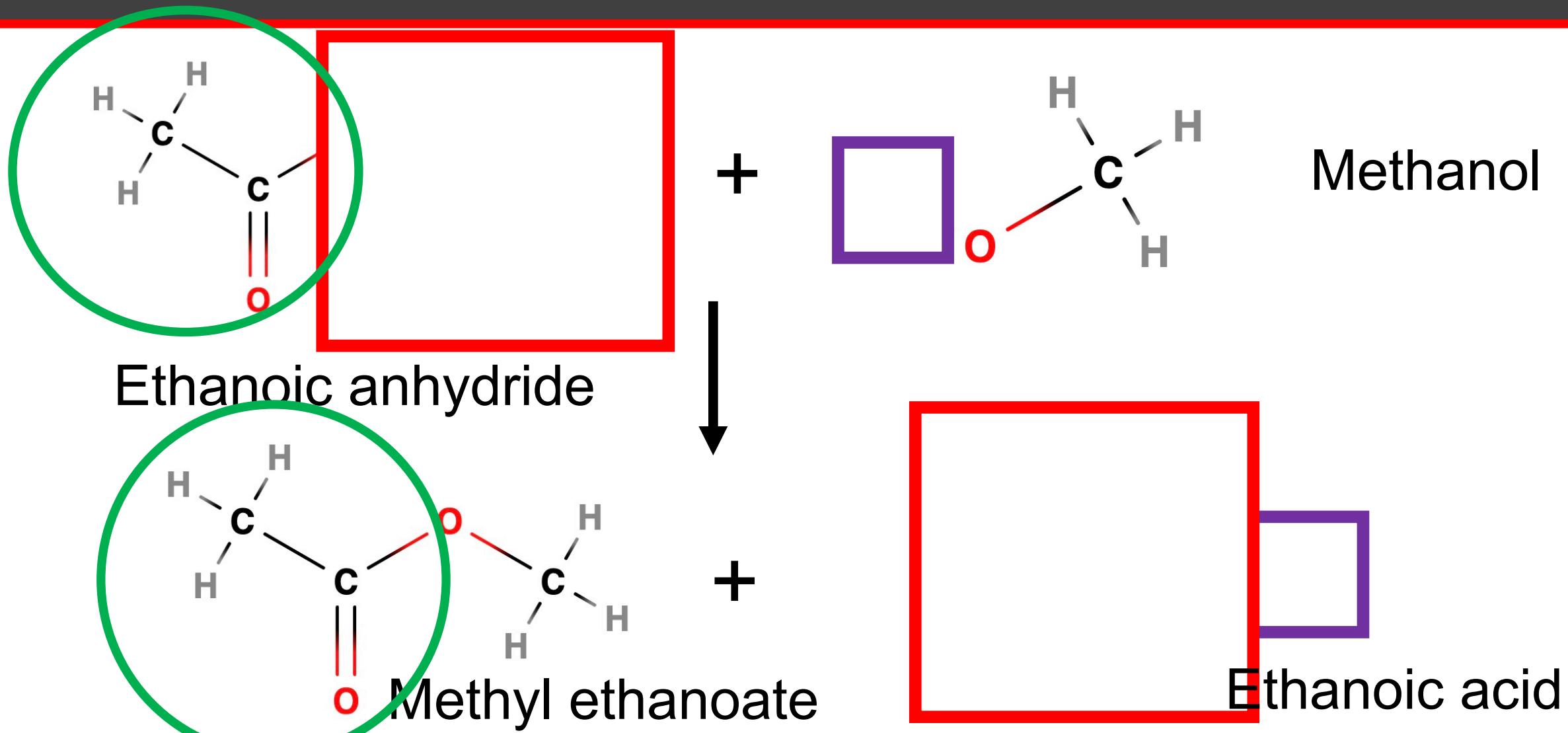
Methyl ethanoate

+

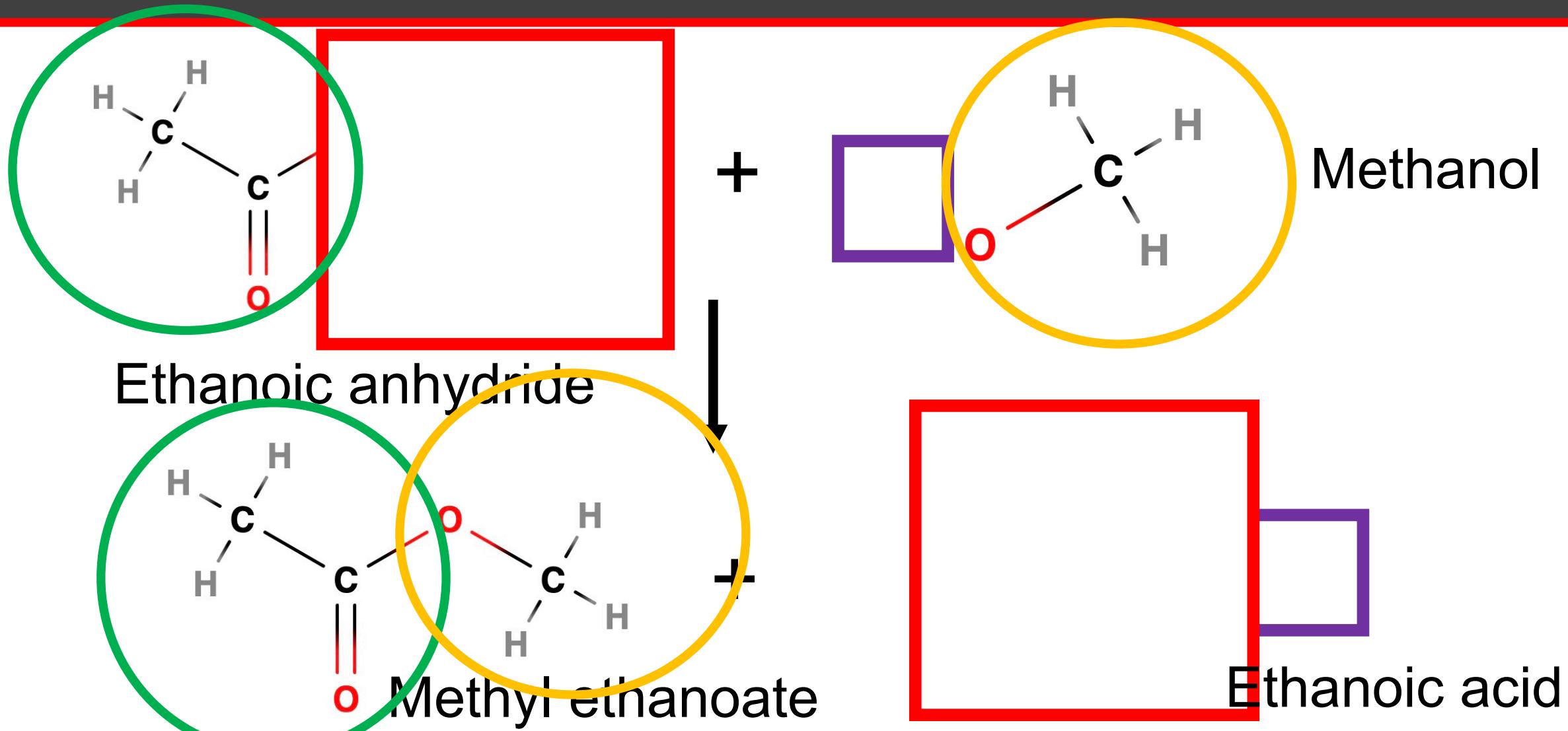


Ethanoic acid

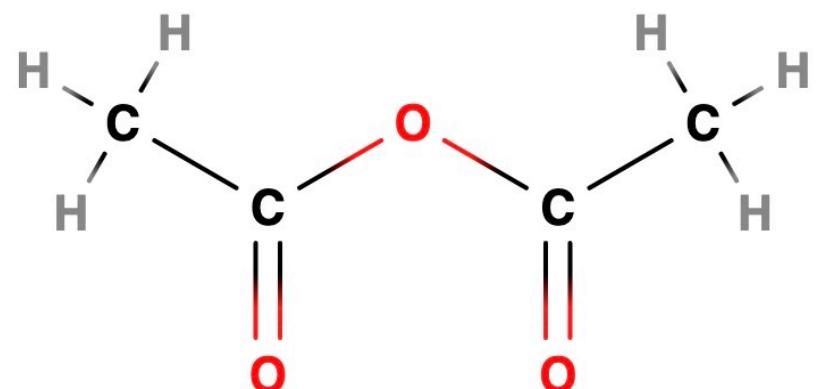
Acid anhydrides in ester formation



Acid anhydrides in ester formation

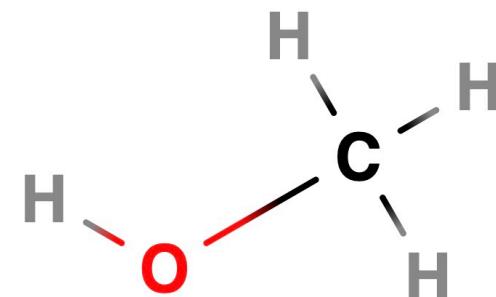


Acid anhydrides in ester formation

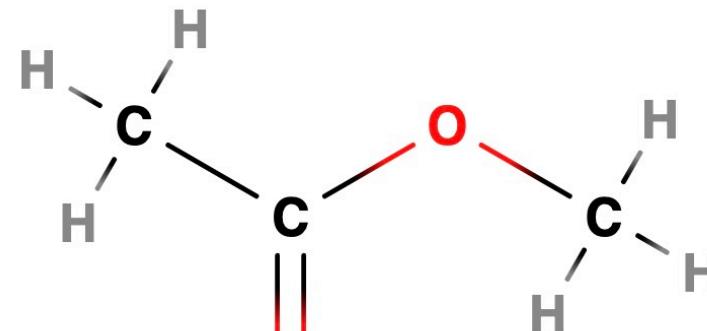


Ethanoic anhydride

+

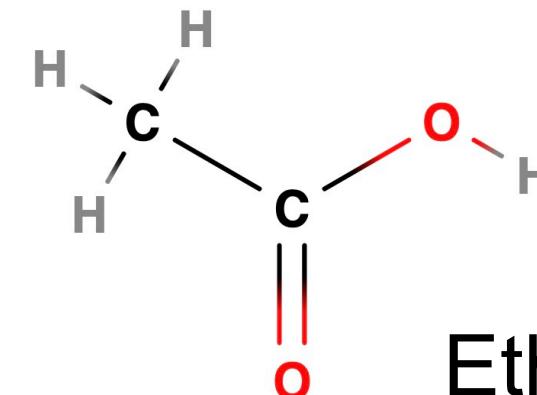


Methanol



Methyl ethanoate

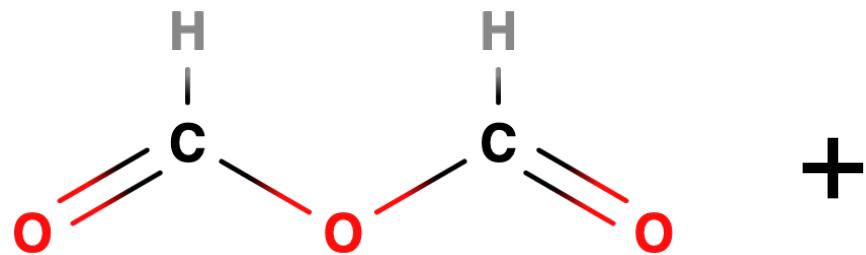
+



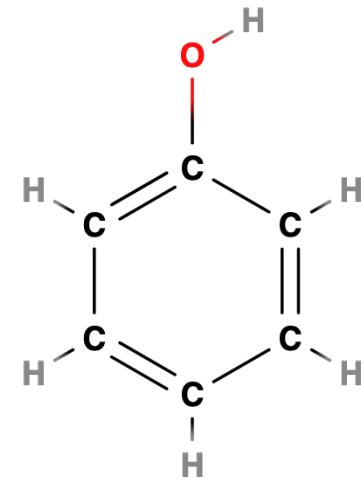
Ethanoic acid

Reaction with phenol

Methanoic
anhydride



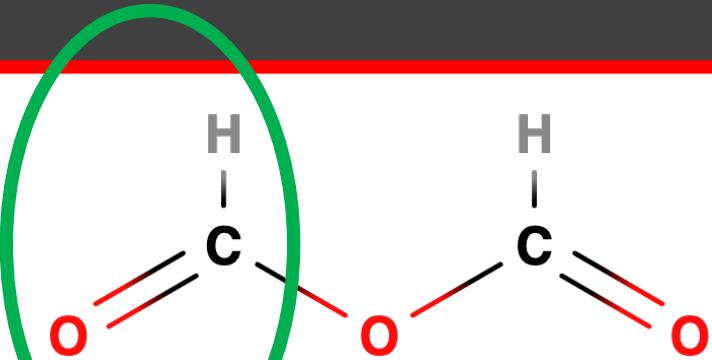
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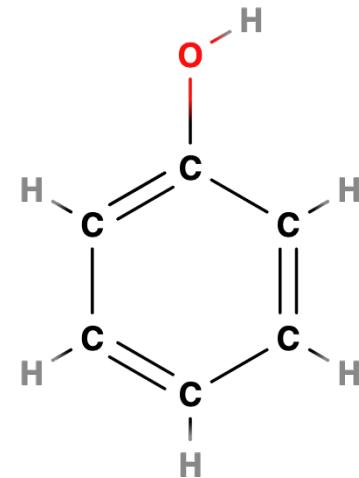
Phenol

Reaction with phenol

Methanoic
anhydride

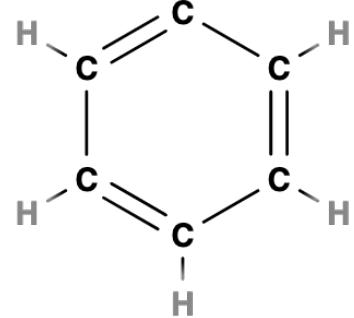


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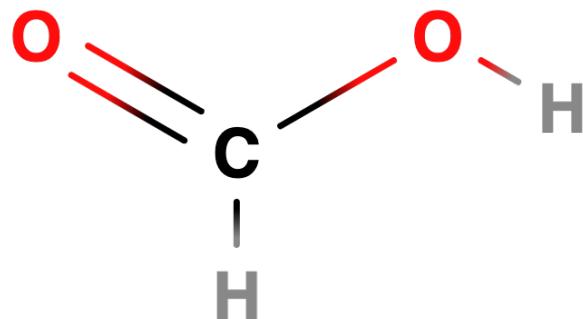


Phenol

Phenyl
methanoate



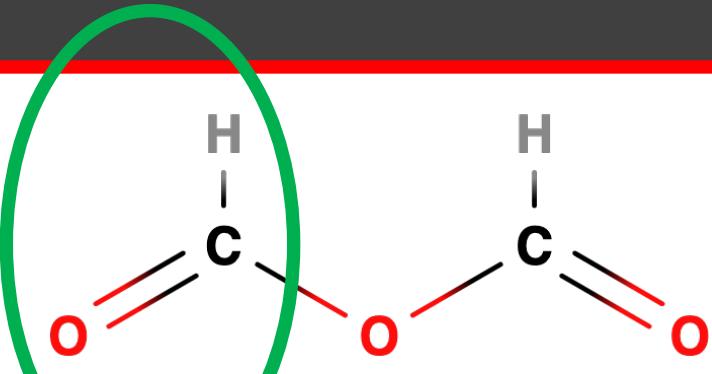
+



Methanol

Reaction with phenol

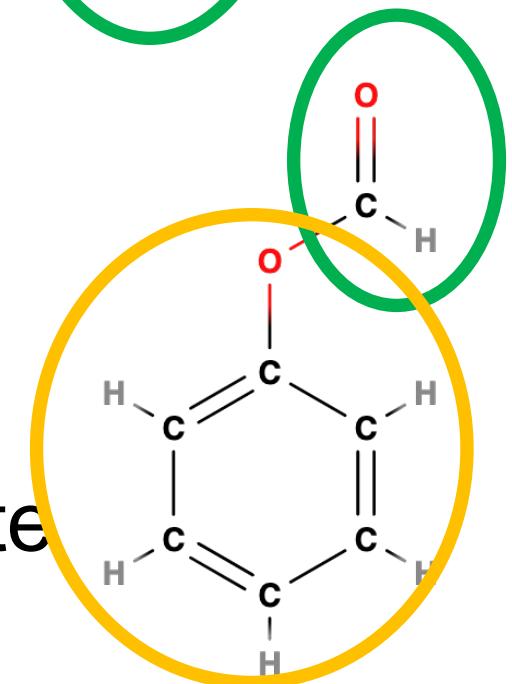
Methanoic
anhydride



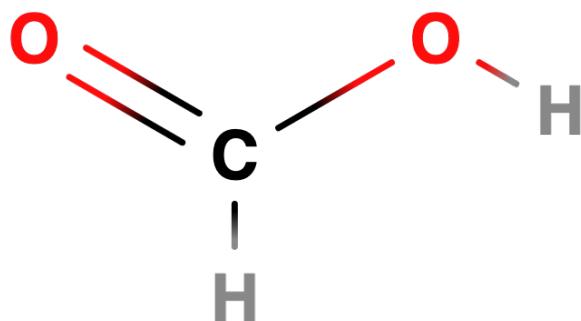
+

Phenol

Phenyl
methanoate



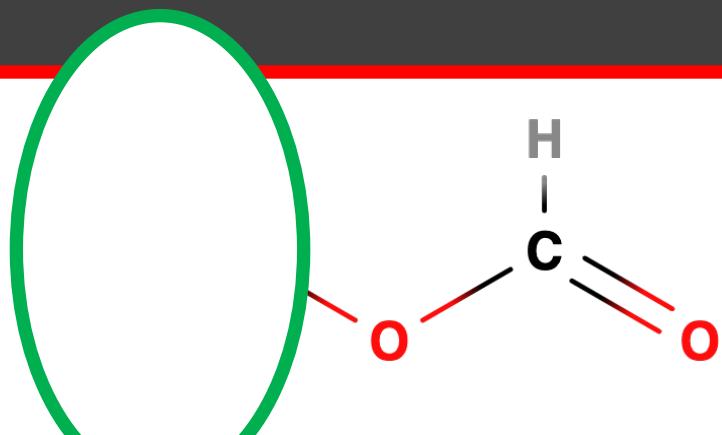
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Methanol

Reaction with phenol

Methanoic
anhydride



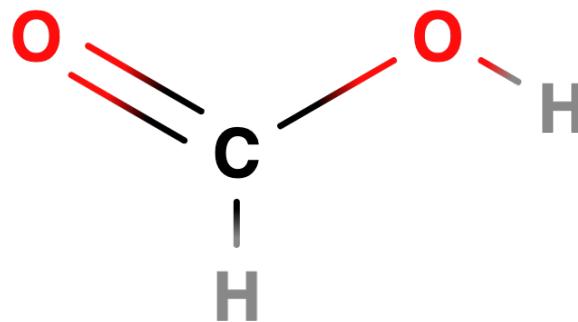
+

Phenol

Phenyl
methanoate



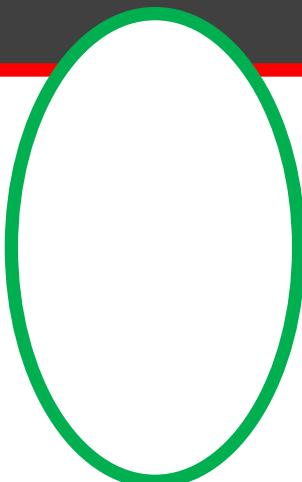
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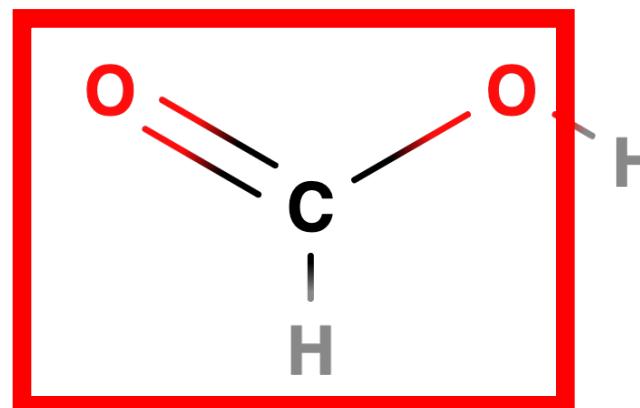
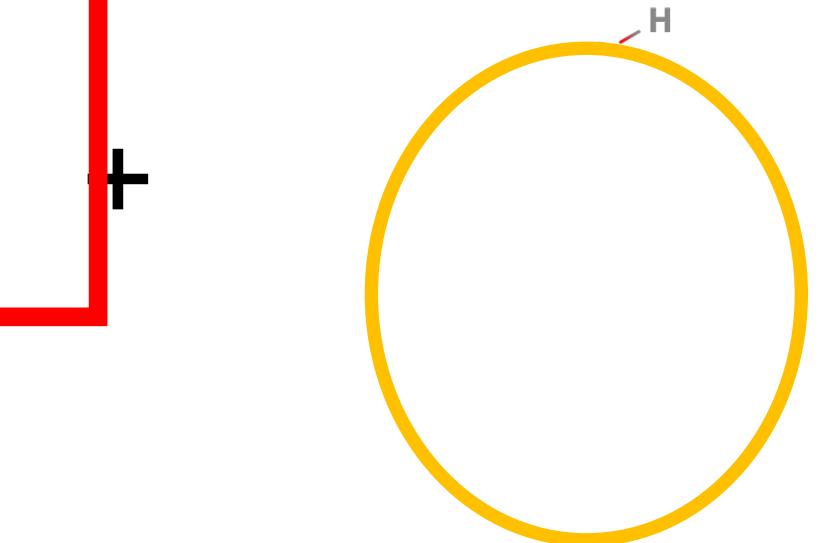
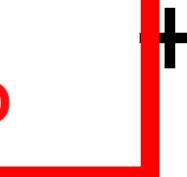
Methanol

Reaction with phenol

Methanoic
anhydride



Phenol

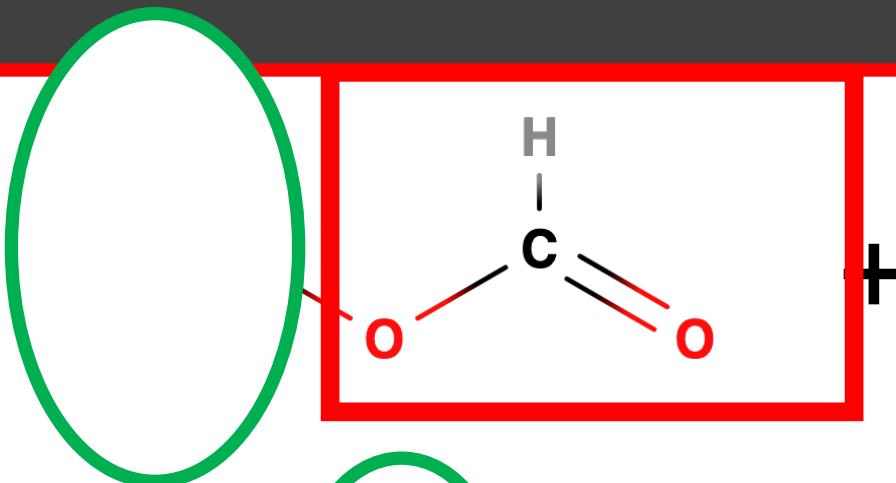


Phenol

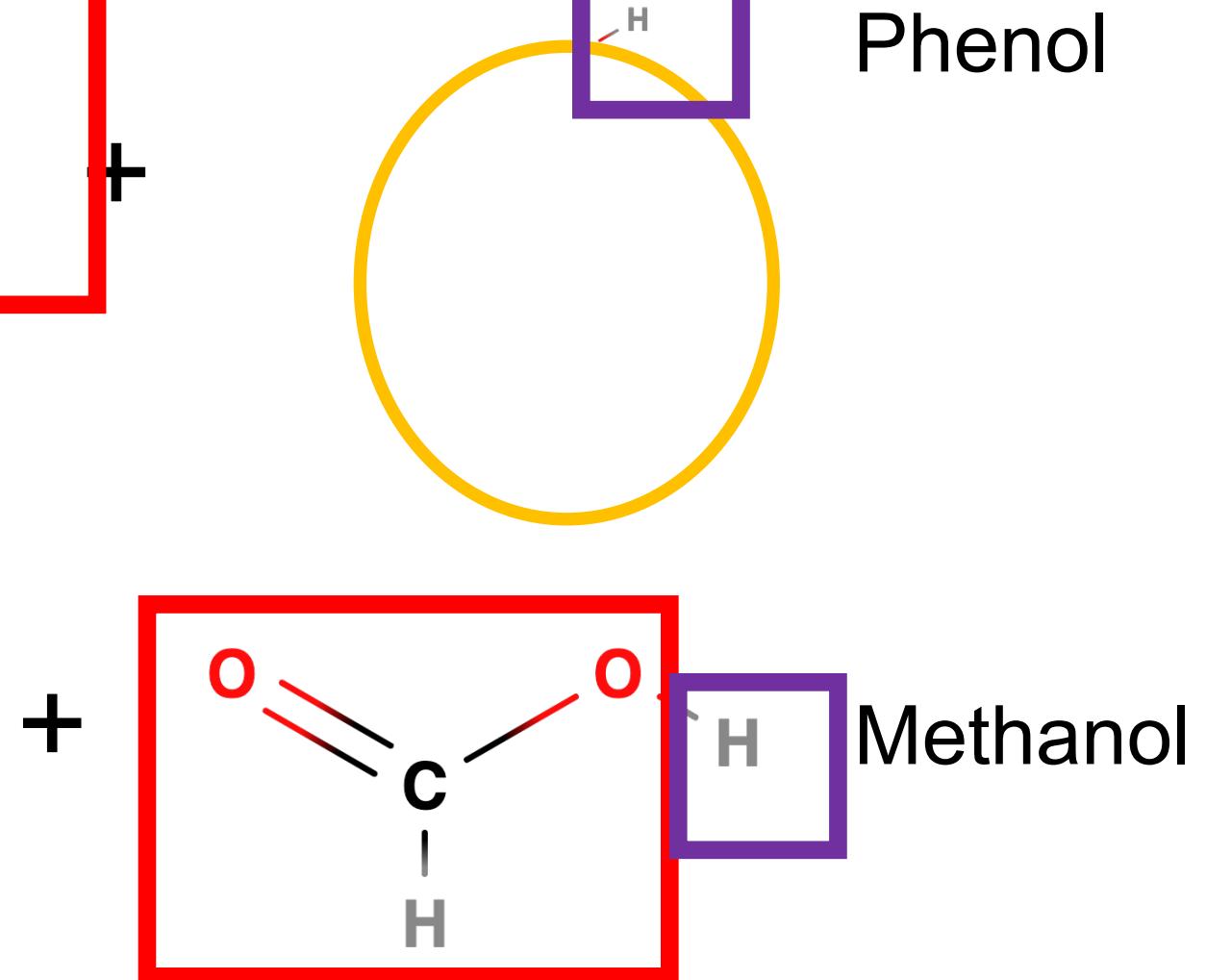
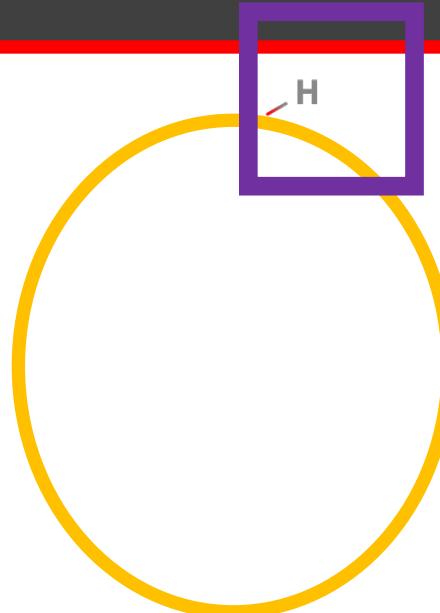
Methanol

Reaction with phenol

Methanoic
anhydride



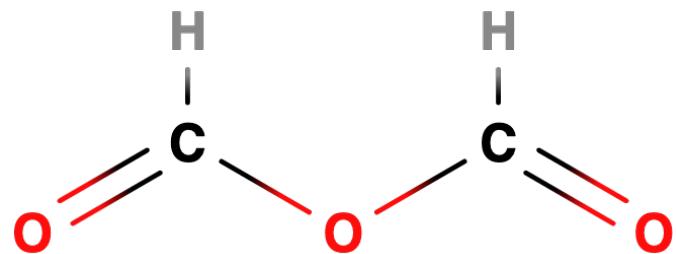
Phenyl
methanoate



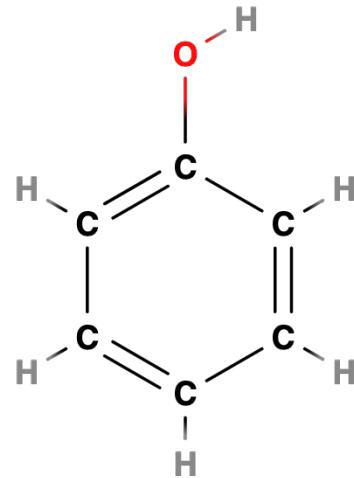
Phenol

Reaction with phenol

Methanoic
anhydride

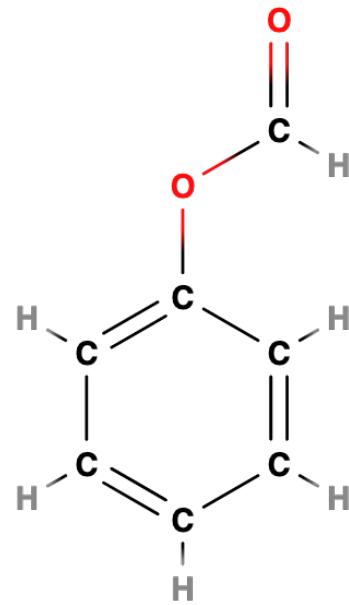


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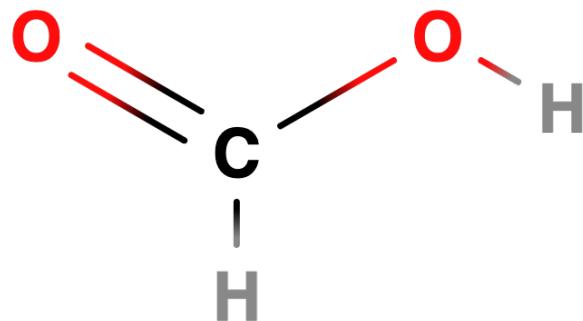


Phenol

Phenyl
methanoate



+



Methanol