

# Professional 1-1 Chemistry Tuition

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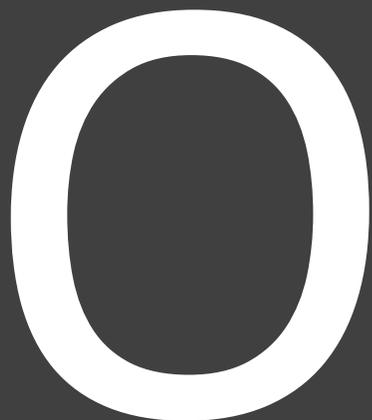
[www.ChemistryTuition.Net](http://www.ChemistryTuition.Net)

Dr Simon Orchard

## Hybridisation and Bonding in Carbon Dioxide

These slides may be downloaded at <https://www.chemistrytuition.net/>

Oxygen A



One pi  
bond, one  
sigma bond  
and two  
lone pairs

Carbon



Two pi  
bonds and  
two sigma  
bonds

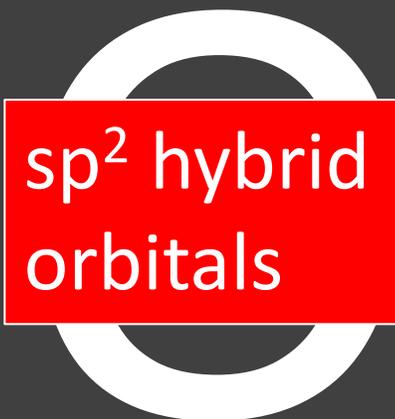
Oxygen B



One pi  
bond, one  
sigma bond  
and two  
lone pairs



Oxygen A



sp<sup>2</sup> hybrid orbitals

One pi bond, one sigma bond and two lone pairs

Carbon



sp hybrid orbitals

Two pi bonds and two sigma bonds

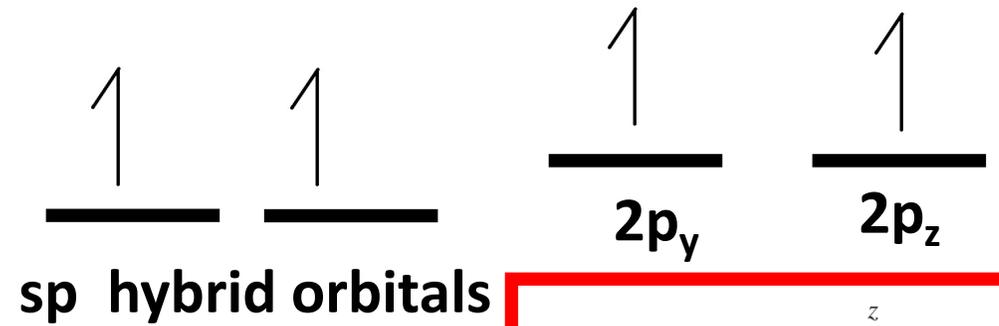
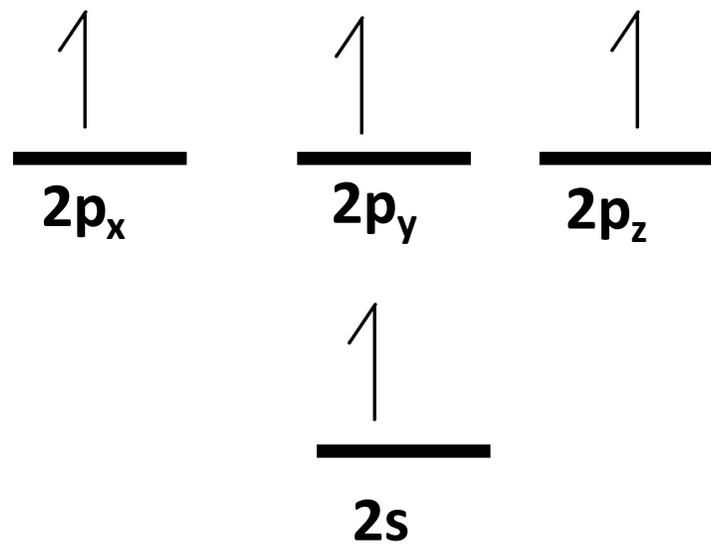
Oxygen B



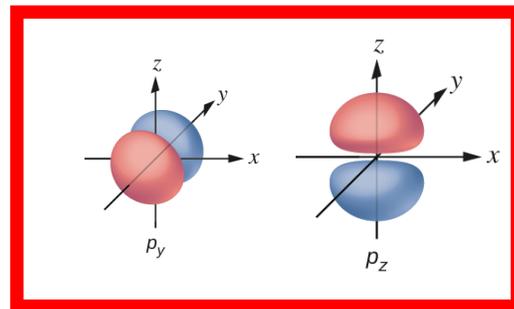
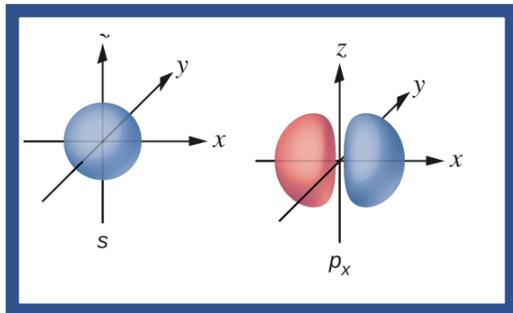
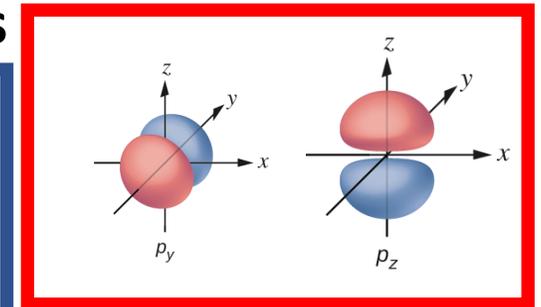
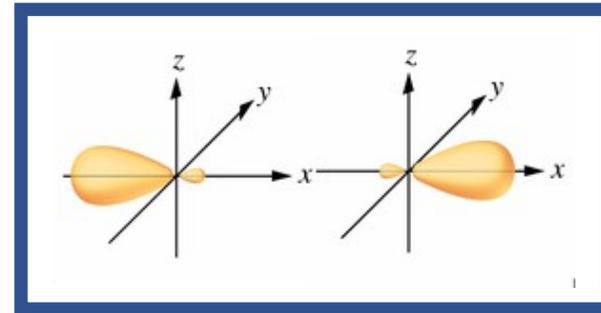
sp<sup>2</sup> hybrid orbitals

One pi bond, one sigma bond and two lone pairs

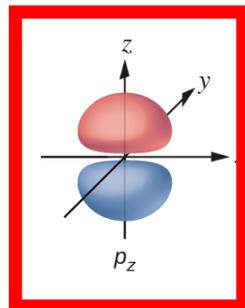
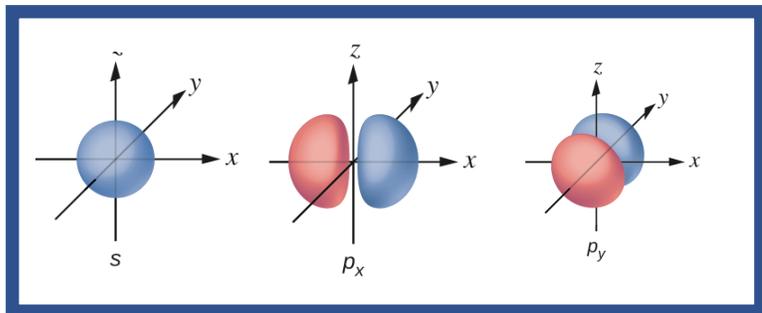
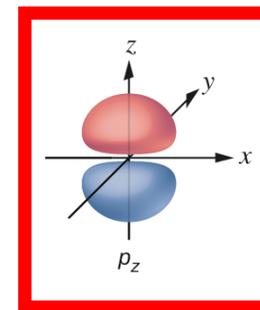
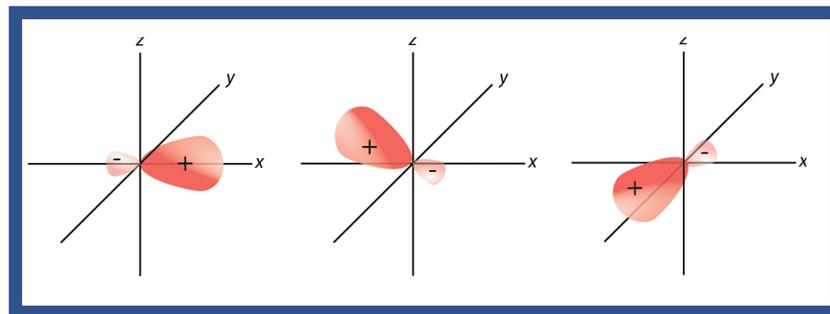
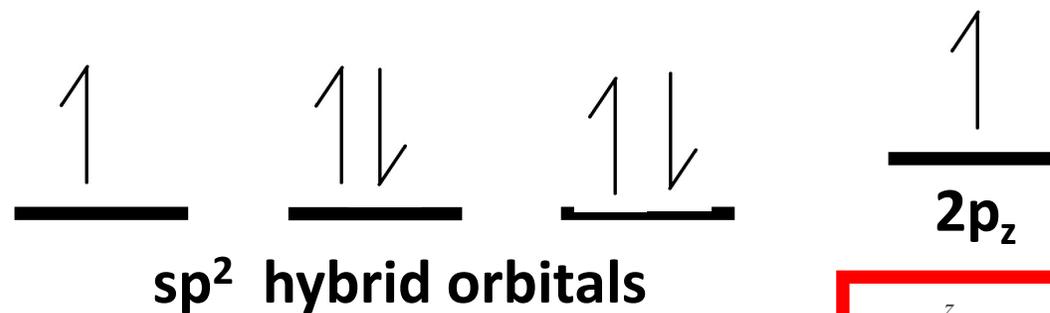
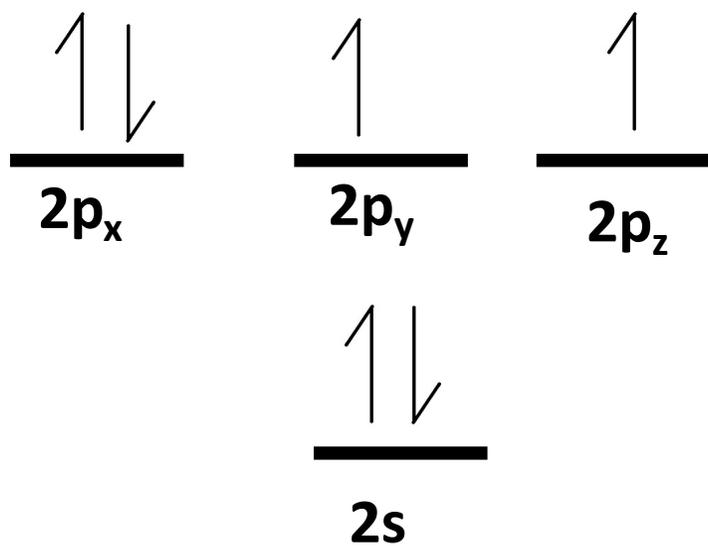
**Carbon – 2s and 2p<sub>x</sub> form sp hybrid orbitals**  
**2p<sub>y</sub> and 2p<sub>z</sub> form pi bonds**



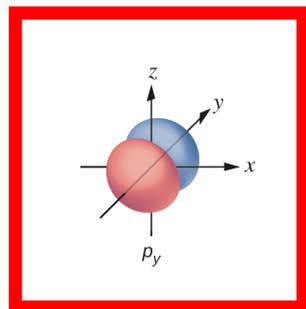
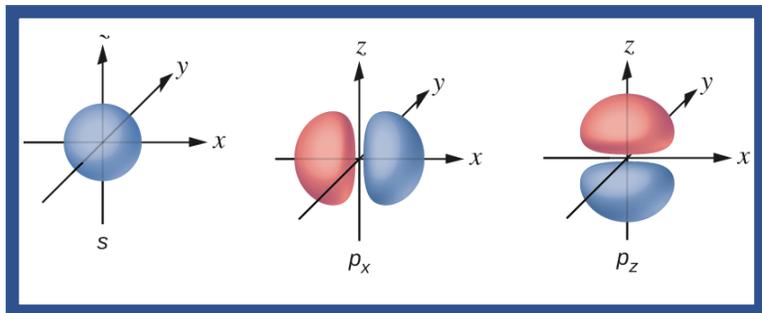
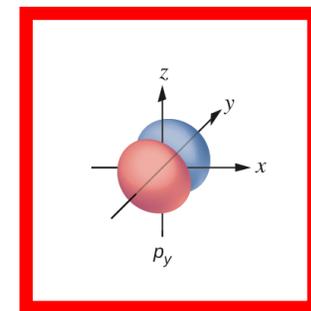
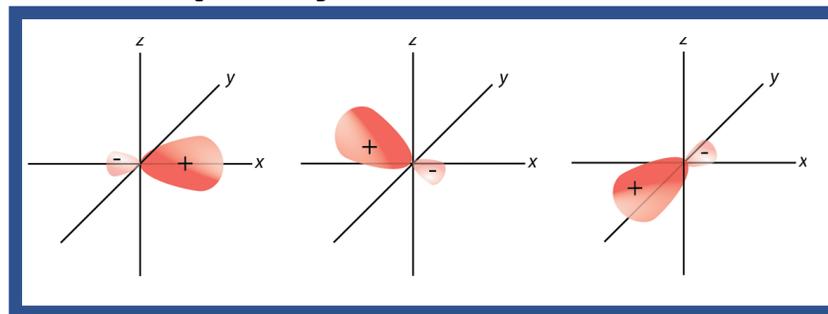
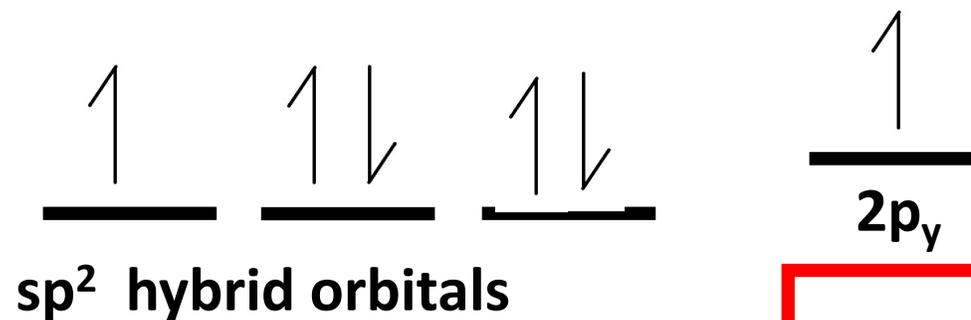
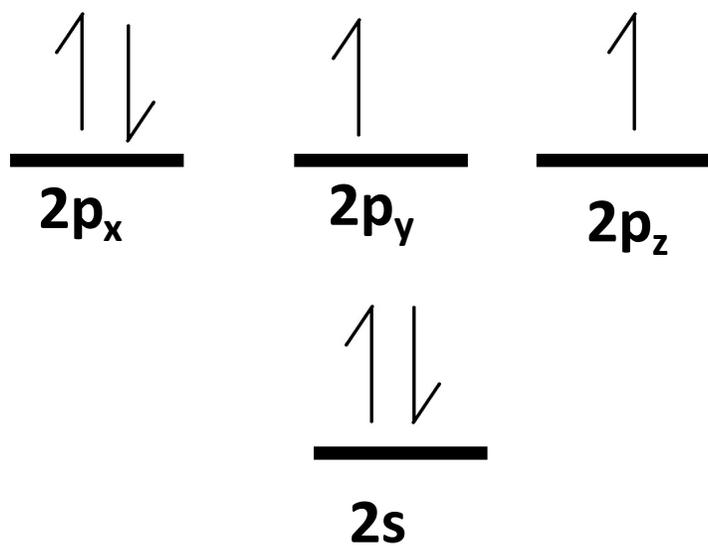
**sp hybrid orbitals**



# Oxygen A – $2s$ , $2p_x$ and $2p_y$ form $sp^2$ hybrid orbitals $2p_z$ forms a pi bond

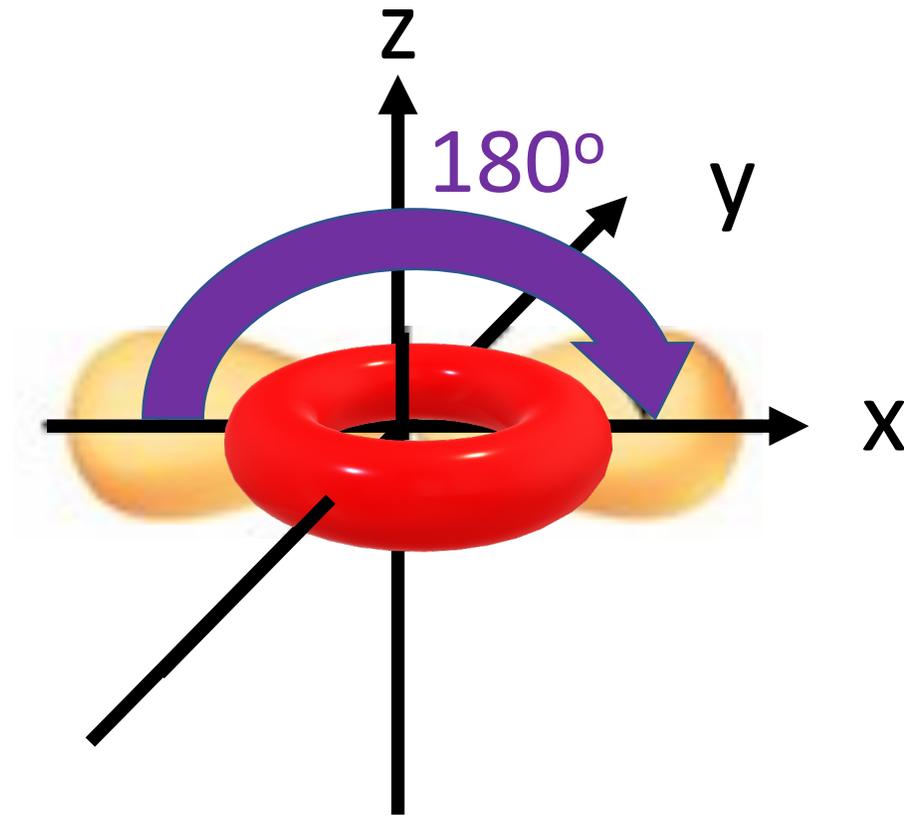


# Oxygen B – $2s$ , $2p_x$ and $2p_z$ form $sp^2$ hybrid orbitals $2p_y$ forms a pi bond



# Sigma Bonding

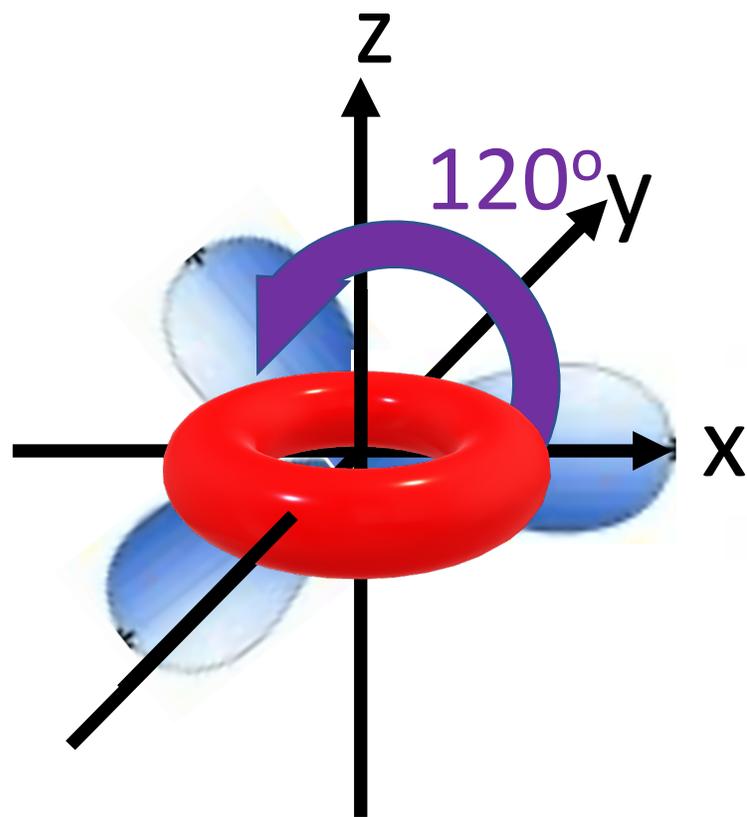
Carbon



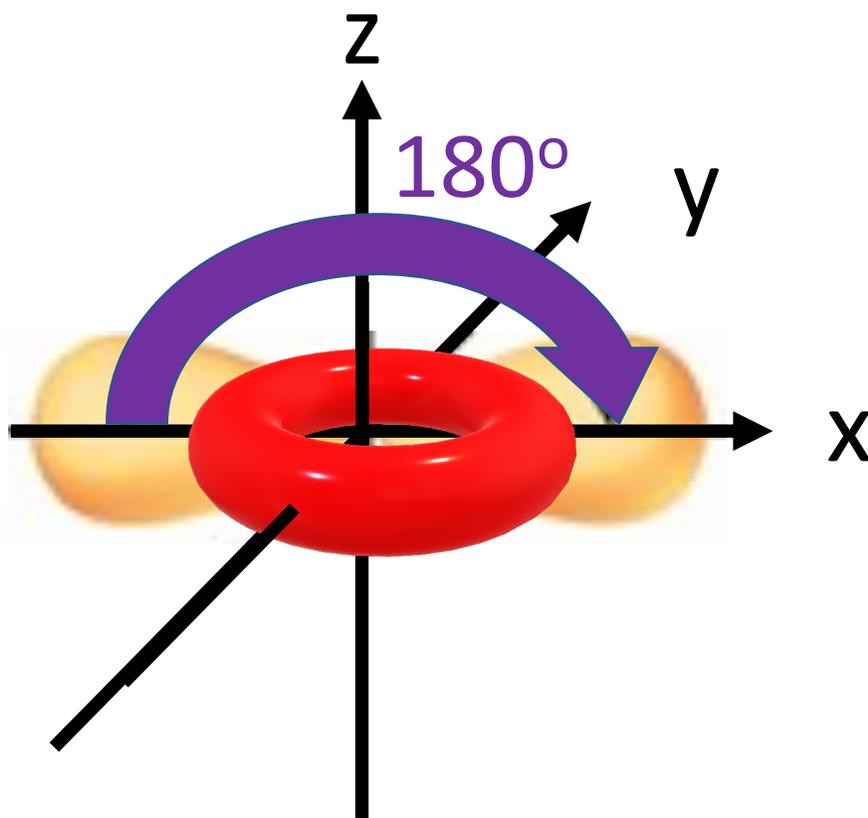
sp hybrid orbitals

$2s, 2p_x$

Oxygen A



Carbon



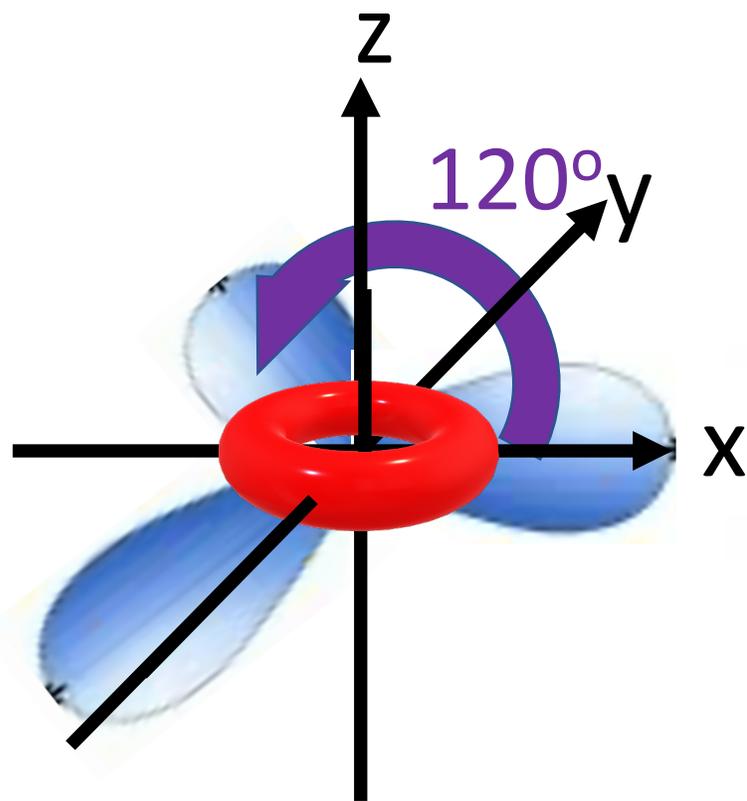
$sp^2$  hybrid orbitals

$2s, 2p_x, 2p_y$

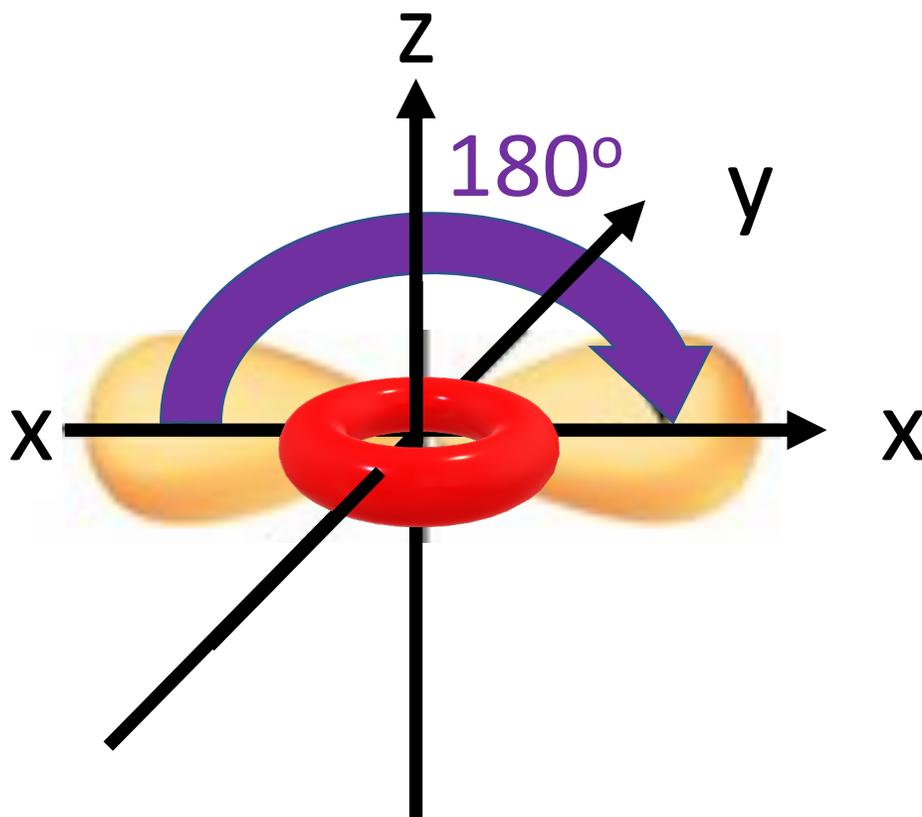
$sp$  hybrid orbitals

$2s, 2p_x$

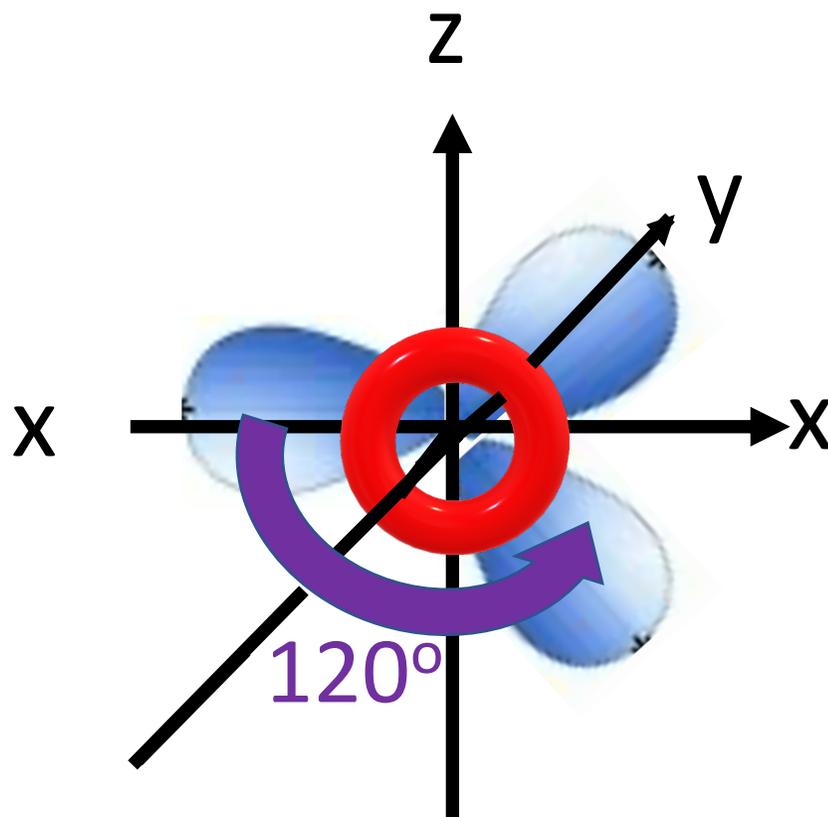
Oxygen A



Carbon



Oxygen B



sp<sup>2</sup> hybrid orbitals

2s, 2p<sub>x</sub>, 2p<sub>y</sub>

sp hybrid orbitals

2s, 2p<sub>x</sub>

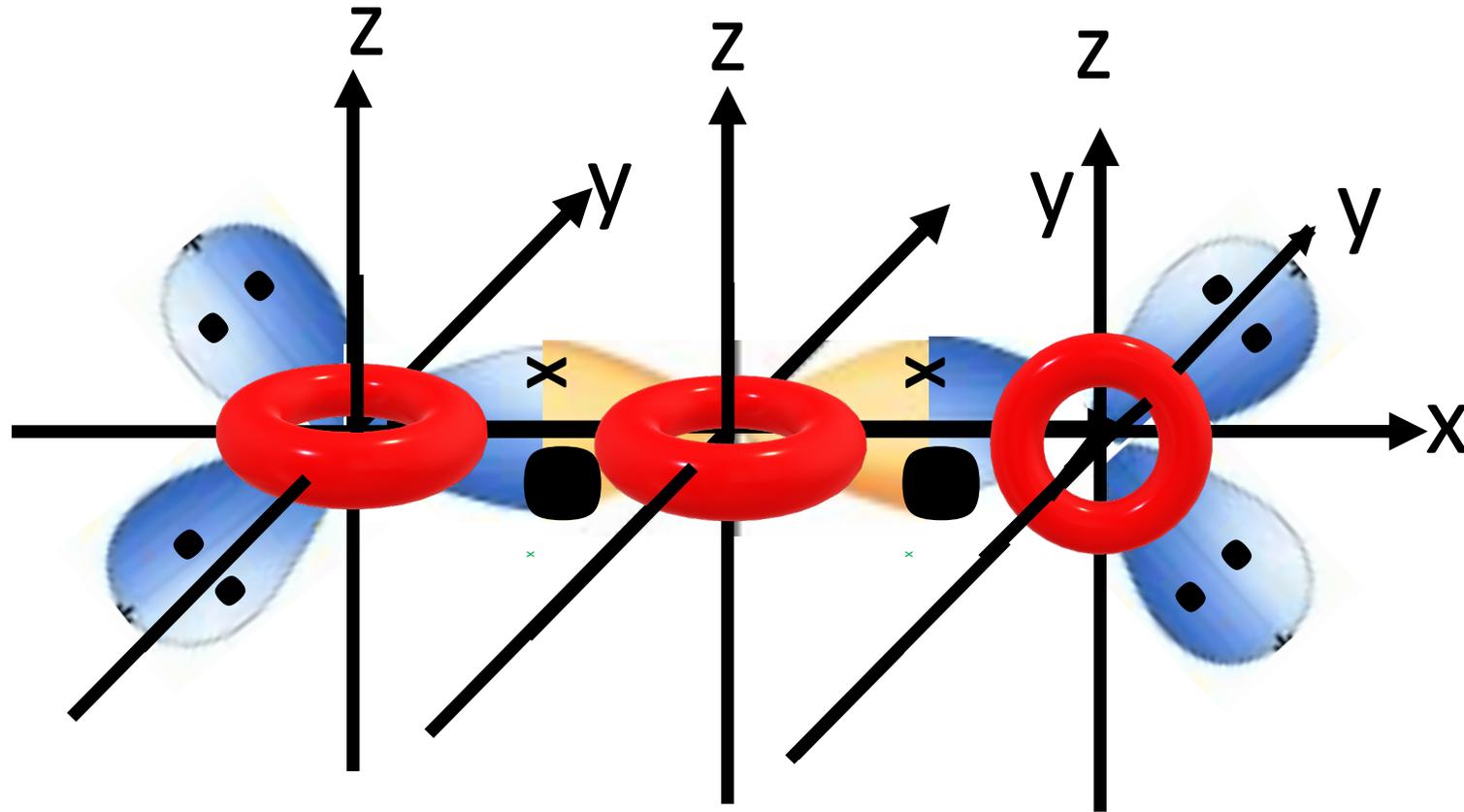
sp<sup>2</sup> hybrid orbitals

2s, 2p<sub>x</sub>, 2p<sub>z</sub>

Oxygen A

Carbon

Oxygen B



$sp^2$  hybrid orbitals

$2s, 2p_x, 2p_y$

$sp$  hybrid orbitals

$2s, 2p_x$

$sp^2$  hybrid orbitals

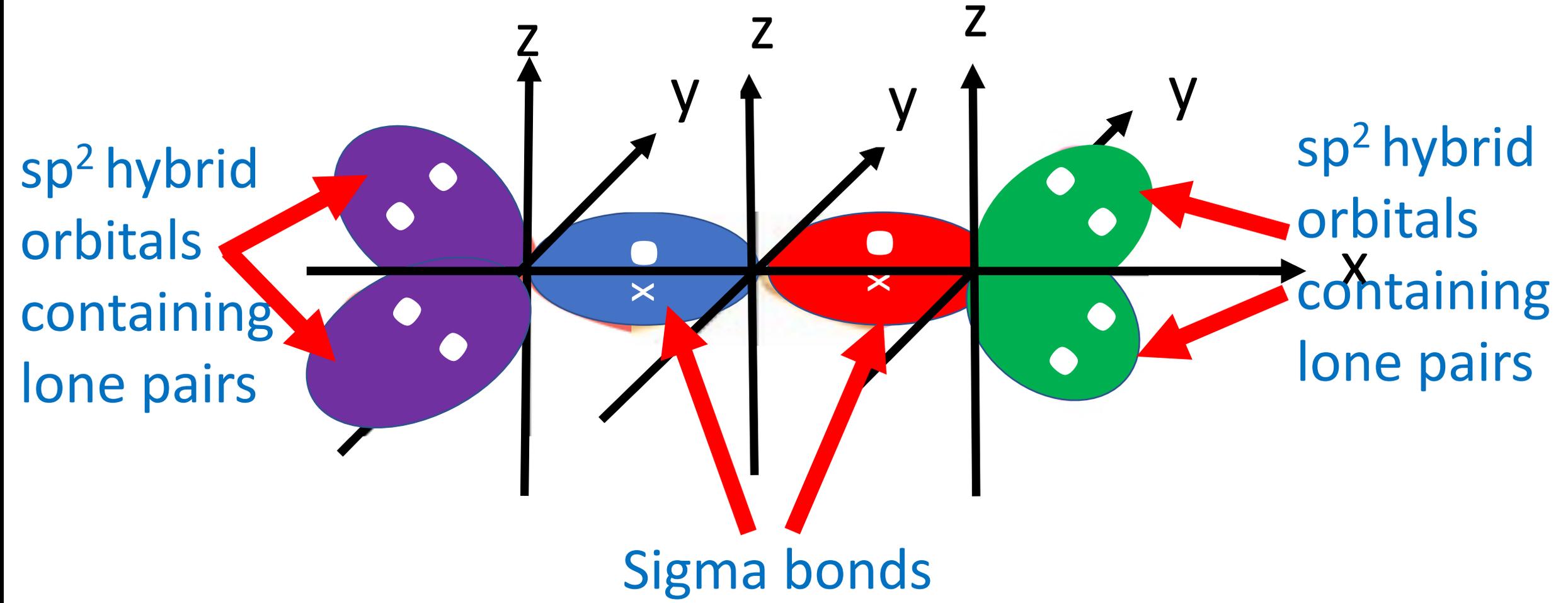
$2s, 2p_x, 2p_z$

# Sigma Framework

Oxygen A

Carbon

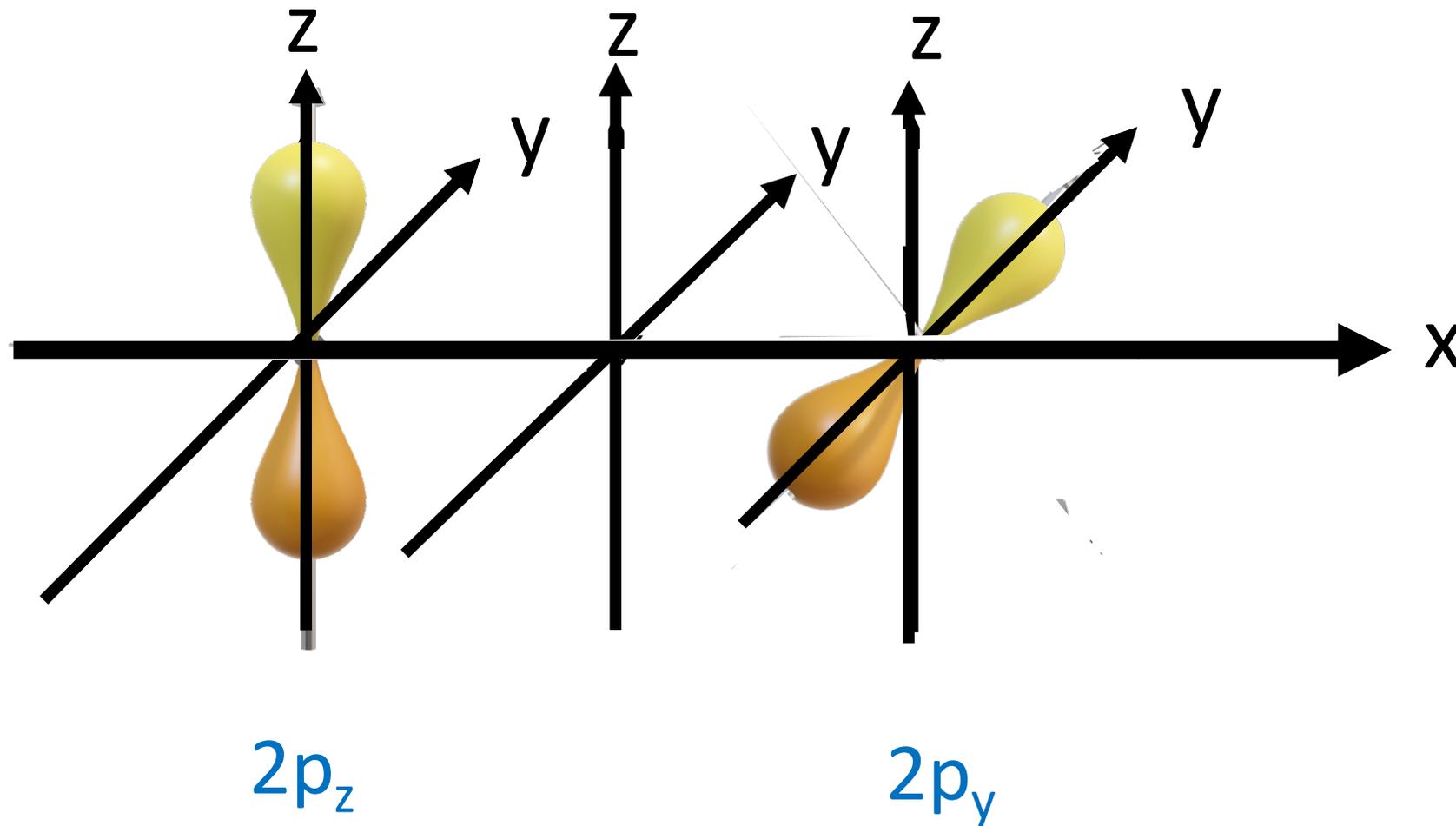
Oxygen B



# Pi Bonding

Oxygen A

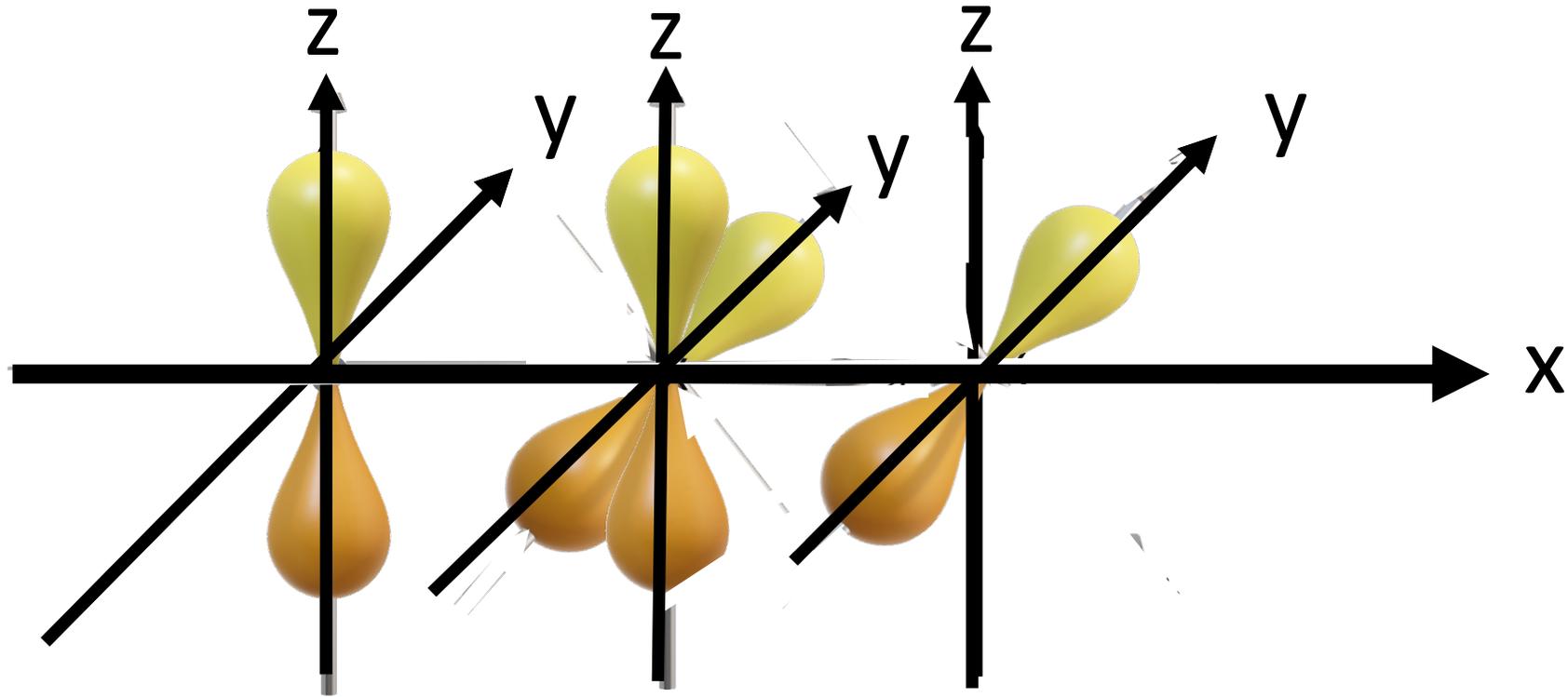
Oxygen B



Oxygen A

Carbon

Oxygen B



$2p_z$

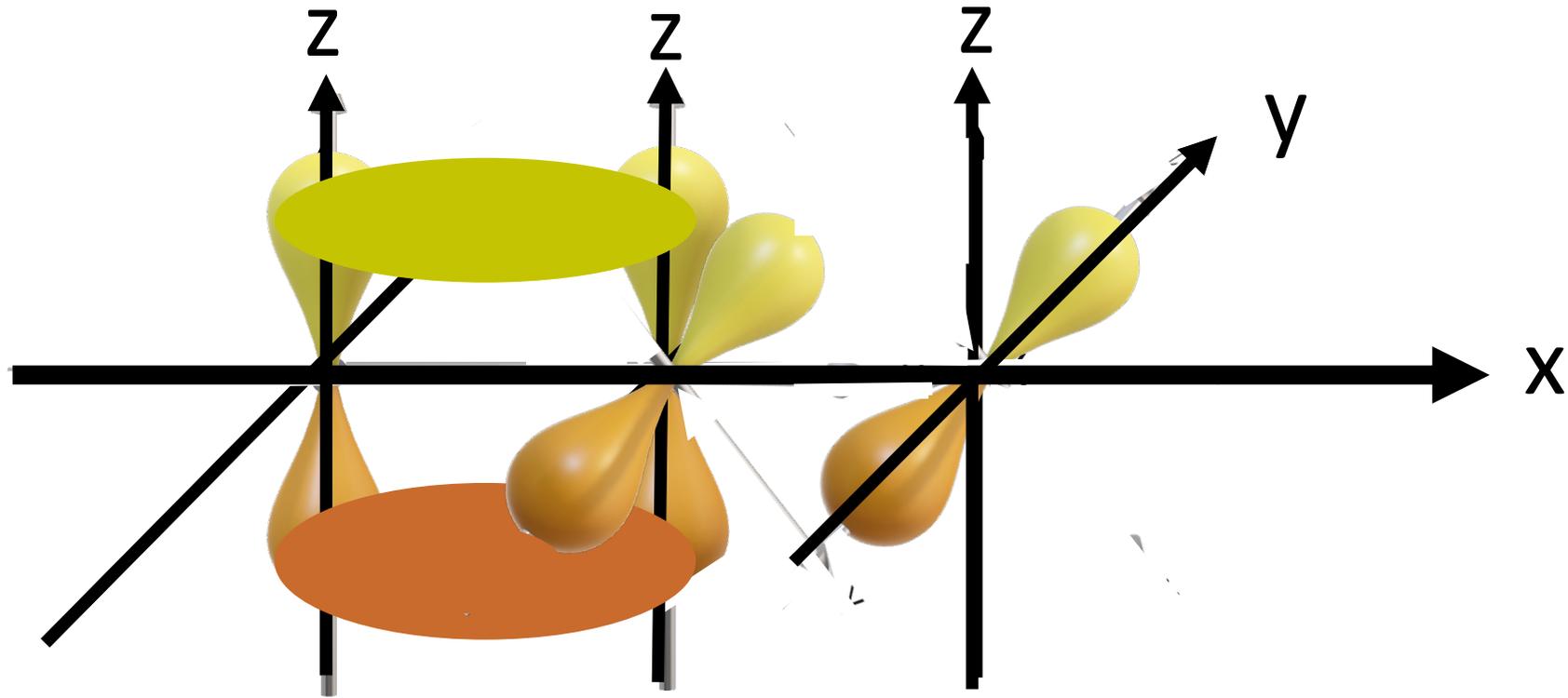
$2p_z, 2p_y$

$2p_y$

Oxygen A

Carbon

Oxygen B



$2p_z$

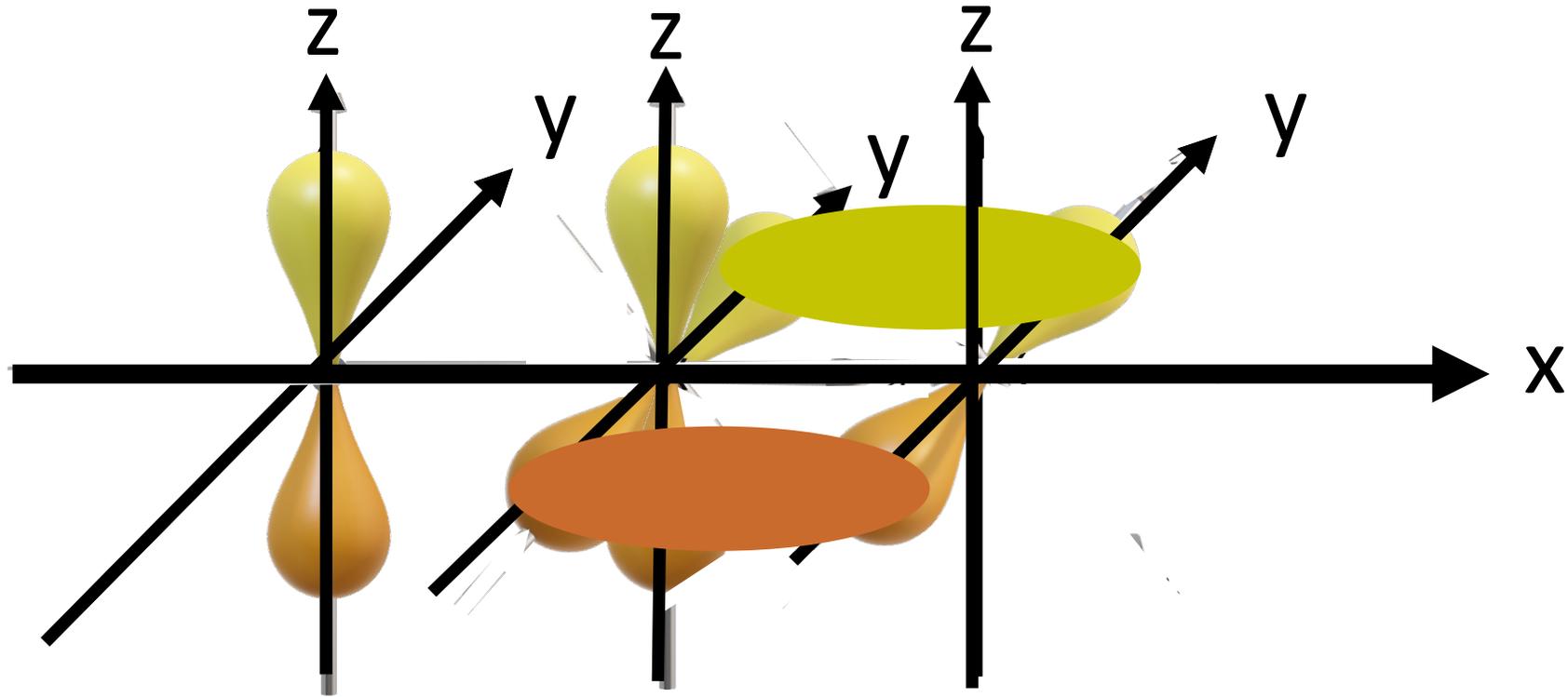
$2p_z, 2p_y$

$2p_y$

Oxygen A

Carbon

Oxygen B



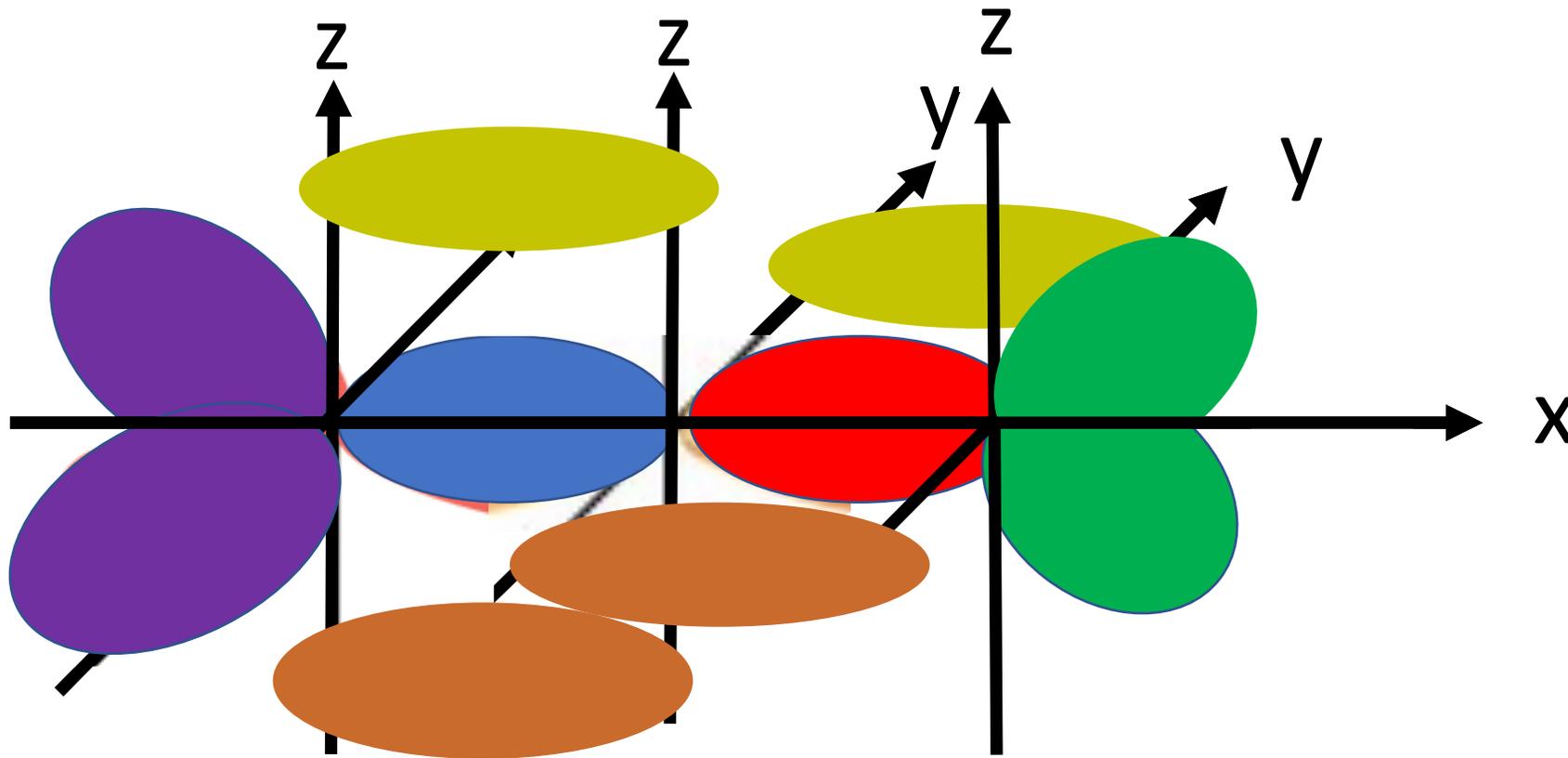
$2p_z$

$2p_z, 2p_y$

$2p_y$

# Overview

Oxygen A      Carbon      Oxygen B



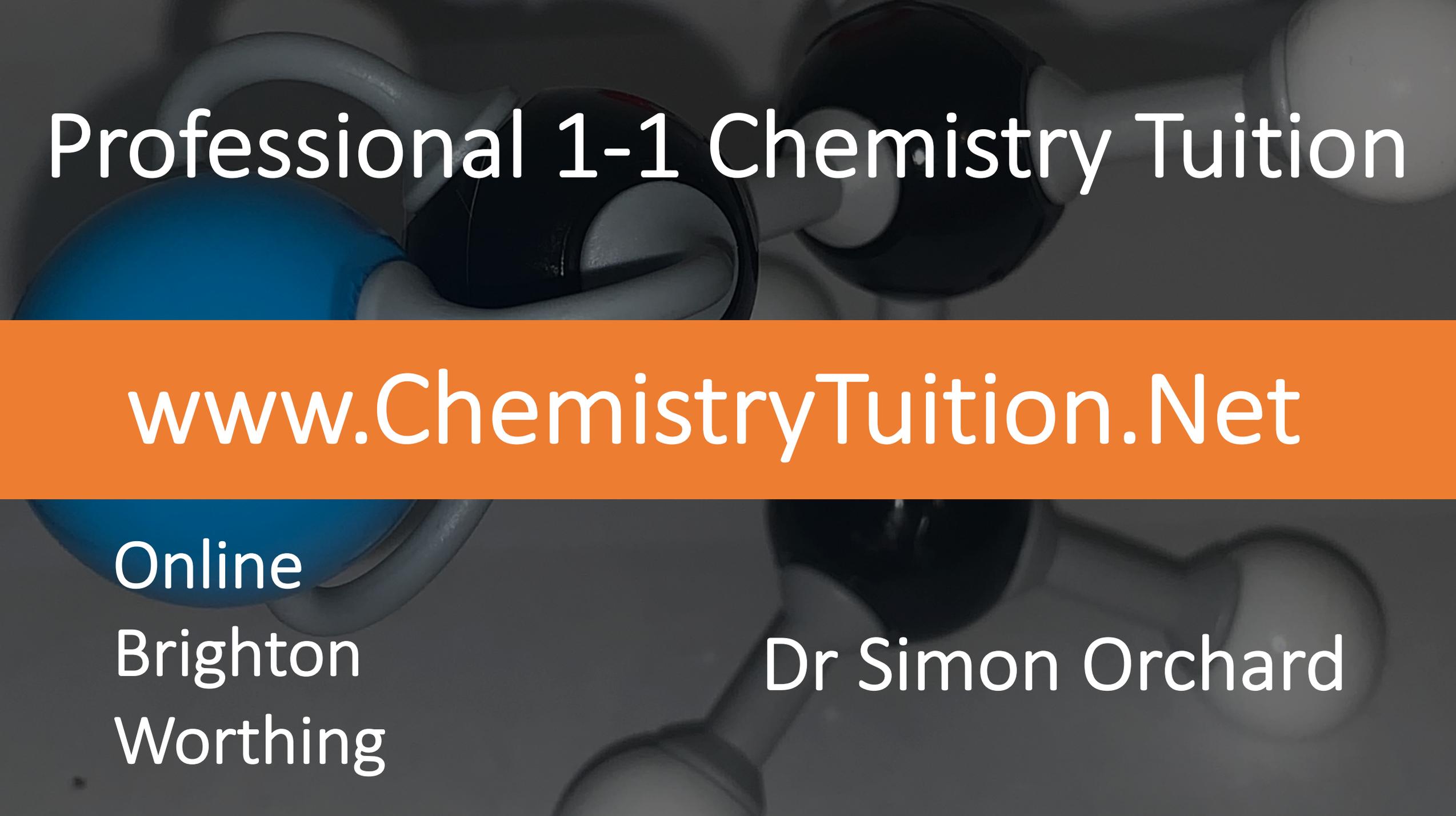
$sp^2$  hybrid orbitals

$2s, 2p_x, 2p_y$

$sp$  hybrid orbitals

$2s, 2p_x$

$2s, 2p_x, 2p_z$



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